# Structure of 2-Methoxy Methyl $\boldsymbol{\beta}$-Orcinol Carboxylate (Methyl 4-Hydroxy-2-methoxy-3,6-dimethylbenzoate) 

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(Received 21 August 1991; accepted 24 October 1991)


#### Abstract

C}_{11} \mathrm{H}_{14} \mathrm{O}_{4}, M_{r}=210.2\), monoclinic, $P 2_{1}, a$ $=7.931$ (1), $\quad b=7.950$ (1), $\quad c=8.929$ (1) $\AA, \quad \beta=$ 102.81 (1) ${ }^{\circ}, V=549.0 \AA^{3}, Z=2, D_{m}$ (flotation in $\left.\mathrm{KI} / \mathrm{H}_{2} \mathrm{O}\right)=1.27, D_{x}=1.272 \mathrm{Mg} \mathrm{m}^{-3}, \lambda(\mathrm{Mo} \mathrm{K} \mathrm{\alpha})=$ $0.71073 \AA, \mu=0.09 \mathrm{~mm}^{-1}, F(000)=224$, room temperature, $R=0.040, w R=0.052$ for 1161 observed reflections. The phenyl ring is inclined to the carboxylate group by 71.7 (1) ${ }^{\circ}$ and to the methoxy group by $75.3(3)^{\circ}$, the latter two groups being inclined to one another by 53.4 (2) $)^{\circ}$. In the crystal, molecules related by the twofold screw axis are linked by a strong intermolecular hydrogen bond involving the 4 -hydroxy substituent and the carboxylate carbonyl group.


Experimental. The title compound (2) is the 2methoxy derivative of the natural product, methyl $\beta$-orcinol carboxylate (1) isolated from 'oak moss' Evernia prunastri (L.) Ach (Tabacchi \& Nicollier, 1977), and was obtained by chemical transformation of (1). First the hydroxyl substituent para to the

(1)

(2)
carboxylate group was protected by benzylation. The hydroxyl group in the ortho position was then methylated using $\mathrm{CH}_{3} \mathrm{I}$ in the presence of $\mathrm{K}_{2} \mathrm{CO}_{3}$ and finally the protecting group was removed by hydrogenolysis (Gunzinger, 1985). Transparent block-like crystals were obtained by slow evaporation of a toluene $/ 2$-propanol $(\approx 1 / 1)$ solution. A crystal of dimensions $0.38 \times 0.30 \times 0.27 \mathrm{~mm}$ was used for data collection using a Stoe AED2 four-circle diffractometer with graphite-monochromated Mo $K \alpha$ radiation. $<2 \%$ intensity variation for 5 standard reflections measured every hour. Accurate cell parameters from $\pm \omega$ values of 20 reflections and

[^0]0108-2701/92/061 126-02\$06.00
their equivalents in the range $25<2 \theta<35^{\circ} .2691$ reflections were measured by the $\omega / 2 \theta$ scan mode with $\theta_{\text {max }}=27.5^{\circ}$, and index limits $h-10$ to 10 , $k-10$ to $10, l 0$ to 11.1361 unique reflections, $R_{\text {int }}=$ $0.026 ; 1161[I>2.0 \sigma(I)]$ were considered observed and used for all further calculations. Systematic absences $0 k 0$ when $k$ odd. Space group $P 2_{1}$ or $P 2_{1} / m$. The $E$ statistics indicated a noncentrosymmetric arrangement and this was confirmed by the successful refinement. Structure solved using SHELXS86 (Sheldrick, 1986). Refinement and all further calculations were carried out using the NRCVAX system (Gabe, Le Page, Charland \& Lee, 1989). In order to fix the origin in space group $P 2_{1}$ the $y$ coordinate of atom $\mathrm{O}(1)$ was held fixed. H atoms located from difference maps and refined isotropically. No attempt was made to determine the absolute structure of the crystal of (2) chosen. Weighted full-matrix least-squares refinement for 1107 reflections gave $R$ $=0.040$ and $w R=0.052$; function minimized $\sum w\left(\left|F_{o}\right|-\left|F_{c}\right|\right)^{2}, \quad w^{-1}=\sigma^{2}\left(F_{o}\right)+0.001\left(F_{o}^{2}\right)$. In the final cycle of refinement the maximum shift $/ \sigma$ ratio was 0.160 . Residual density limits in final difference map +0.14 and -0.18 e $\AA^{-3}$. Neutral complexatom scattering factors in $N R C V A X$ from International Tables for X-ray Crystallography (1974, Vol. IV). Final positional $\dagger$ and equivalent isotropic thermal parameters are given in Table 1 and interatomic distances and angles in Table 2. The numbering scheme is illustrated in Fig. 1.

Related literature. In the title compound (2) the orientation of the carboxylate group and the adjacent methoxy group with respect to the phenyl ring is very similar to that found in $(R, R, R)$-methyl 2,3-dimethoxy-4-hydroxy-5-iodo-6-methylbenzoate (Nicolaou, Ebata, Stylianides, Groneberg \& Carrol, 1988). The intermolecular hydrogen bonding involv-

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Table 1. Atomic parameters $x, y, z$ and $B_{\text {iso }}$ values E.s.d.'s refer to the last digit printed.

|  | $x$ | $y$ | $z$ | $B_{\text {iso }} *\left(\AA^{2}\right)$ |
| :--- | :---: | :---: | :---: | :---: |
|  |  |  |  |  |
| $\mathrm{O}(1)$ | $0.6623(3)$ | 0.4171 | $0.6727(2)$ | $4.92(9)$ |
| $\mathrm{O}(2)$ | $0.4723(3)$ | $0.2218(4)$ | $0.7017(2)$ | $4.74(8)$ |
| $\mathrm{O}(3)$ | $0.5325(2)$ | $0.3685(4)$ | $1.0185(2)$ | $3.96(8)$ |
| $\mathrm{O}(4)$ | $1.0236(3)$ | $0.0825(4)$ | $1.3094(2)$ | $4.92(9)$ |
| $\mathrm{C}(1)$ | $0.7289(3)$ | $0.2359(4)$ | $0.8916(3)$ | $3.10(8)$ |
| $\mathrm{C}(2)$ | $0.6787(3)$ | $0.2708(4)$ | $1.0283(3)$ | $3.11(9)$ |
| $\mathrm{C}(3)$ | $0.7780(3)$ | $0.2234(4)$ | $1.1701(3)$ | $3.37(9)$ |
| $\mathrm{C}(4)$ | $0.9281(3)$ | $0.1314(5)$ | $1.1701(3)$ | $3.46(9)$ |
| $\mathrm{C}(5)$ | $0.9786(3)$ | $0.0949(5)$ | $1.0348(3)$ | $3.45(9)$ |
| $\mathrm{C}(6)$ | $0.8819(3)$ | $0.1485(4)$ | $0.8931(3)$ | $3.30(9)$ |
| $\mathrm{C}(7)$ | $0.6204(3)$ | $0.3034(5)$ | $0.7447(3)$ | $3.33(9)$ |
| $\mathrm{C}(8)$ | $0.3530(5)$ | $0.2827(7)$ | $0.5661(4)$ | $5.63(16)$ |
| $\mathrm{C}(9)$ | $0.9394(5)$ | $0.1055(6)$ | $0.7479(4)$ | $5.16(14)$ |
| $\mathrm{C}(10)$ | $0.7305(5)$ | $0.2734(6)$ | $1.3182(4)$ | $4.83(14)$ |
| $\mathrm{C}(11)$ | $0.3825(4)$ | $0.2798(6)$ | $1.0400(5)$ | $5.13(15)$ |

* $B_{\text {iso }}$ is the mean of the principal axes of the thermal ellipsoid.


Fig. 1. PLUTO (Motherwell \& Clegg, 1978) plot of (2) showing the numbering scheme.
ing the 4-hydroxy substituent and the carboxylate carbonyl group (see Table 2) is also similar to that found in the above-mentioned compound (Nicolaou et al., 1988), and to that found in acetylcholine $\beta$-resorcylate (Jensen, 1975) and (1) (Brehm, Stoeckli-Evans, Tabacchi \& Bürgi, 1983).

Table 2. Final distances $(\AA)$ and angles $\left({ }^{\circ}\right)$

| $\mathrm{O}(1)-\mathrm{C}(7)$ | $1.197(4)$ | $\mathrm{C}(1)-\mathrm{C}(7)$ | $1.500(4)$ |
| :--- | :--- | :--- | :--- |
| $\mathrm{O}(2)-\mathrm{C}(7)$ | $1.322(3)$ | $\mathrm{C}(2)-\mathrm{C}(3)$ | $1.387(4)$ |
| $\mathrm{O}(2)-\mathrm{C}(8)$ | $1.445(4)$ | $\mathrm{C}(3)-\mathrm{C}(4)$ | $1.397(4)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)$ | $1.382(3)$ | $\mathrm{C}(3)-\mathrm{C}(10)$ | $1.507(4)$ |
| $\mathrm{O}(3)-\mathrm{C}(11)$ | $1.432(4)$ | $\mathrm{C}(4)-\mathrm{C}(5)$ | $1.385(4)$ |
| $\mathrm{O}(4)-\mathrm{C}(4)$ | $1.361(3)$ | $\mathrm{C}(5)-\mathrm{C}(6)$ | $1.393(4)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)$ | $1.393(4)$ | $\mathrm{C}(6)-\mathrm{C}(9)$ | $1.506(4)$ |
| $\mathrm{C}(1)-\mathrm{C}(6)$ | $1.396(4)$ | $\mathrm{O}(4)-\mathrm{H}(\mathrm{O} 4)$ | $0.91(6)$ |
| $\mathrm{O}(4) \cdots \mathrm{O}\left(1^{1}\right)$ | $2.790(3)$ | $\mathrm{O}\left(1^{\prime}\right) \cdots \mathrm{H}(\mathrm{O} 4)$ | $1.92(6)$ |
|  |  |  |  |
| $\mathrm{C}(7)-\mathrm{O}(2)-\mathrm{C}(8)$ | $117.0(3)$ | $\mathrm{O}(4)-\mathrm{C}(4)-\mathrm{C}(3)$ | $116.8(3)$ |
| $\mathrm{C}(2)-\mathrm{O}(3)-\mathrm{C}(11)$ | $115.2(3)$ | $\mathrm{O}(4)-\mathrm{C}(4)-\mathrm{C}(5)$ | $121.8(3)$ |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(6)$ | $120.5(2)$ | $\mathrm{C}(3)-\mathrm{C}(4)-\mathrm{C}(5)$ | $121.4(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(1)-\mathrm{C}(7)$ | $118.3(2)$ | $\mathrm{C}(4)-\mathrm{C}(5)-\mathrm{C}(6)$ | $121.4(3)$ |
| $\mathrm{C}(6)-\mathrm{C}(1)-\mathrm{C}(7)$ | $121.0(2)$ | $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(5)$ | $117.6(2)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(1)$ | $117.1(2)$ | $\mathrm{C}(1)-\mathrm{C}(6)-\mathrm{C}(9)$ | $122.3(2)$ |
| $\mathrm{O}(3)-\mathrm{C}(2)-\mathrm{C}(3)$ | $120.6(2)$ | $\mathrm{C}(5)-\mathrm{C}(6)-\mathrm{C}(9)$ | $120.1(3)$ |
| $\mathrm{C}(1)-\mathrm{C}(2)-\mathrm{C}(3)$ | $122.0(2)$ | $\mathrm{O}(1)-\mathrm{C}(7)-\mathrm{O}(2)$ | $123.3(3)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(4)$ | $117.0(2)$ | $\mathrm{O}(1)-\mathrm{C}(7)-\mathrm{C}(1)$ | $124.5(2)$ |
| $\mathrm{C}(2)-\mathrm{C}(3)-\mathrm{C}(10)$ | $121.9(3)$ | $\mathrm{O}(2)-\mathrm{C}(7)-\mathrm{C}(1)$ | $112.3(3)$ |
| $\mathrm{C}(4)-\mathrm{C}(3)-\mathrm{C}(10)$ | $121.1(3)$ | $\mathrm{O}(4)-\mathrm{H}(\mathrm{O} 4) \cdots \mathrm{O}\left(1^{\prime}\right)$ | $159(5)$ |

Symmetry operation: (i) $2-x,-0.5+y, 2-z$.

We wish to thank the Swiss National Science Foundation for financial support.

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Acta Cryst. (1992). C48, 1127-1129

# Structure of 6,7,8,9,10,11-Hexafluoro-2,4-diphenyl-3H-naphtho[2,1-b]azepine 

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(Received 9 July 1991; accepted 15 October 1991)

> Abstract. $\mathrm{C}_{26} \mathrm{H}_{13} \mathrm{~F}_{6} \mathrm{~N}, M_{r}=453.39$, triclinic, $P \overline{\mathrm{I}}, a=$ 13.59 (1), $b=10.62$ (1), $c=7.59$ (1) $\AA, \alpha=110.0$ (1), $\beta=106.2(1), \quad \gamma=91.4(1)^{\circ}, \quad V=979.74 \AA^{3}, \quad Z=2$,

> 0108-2701/92/061127-03\$06.00
> $D_{x}=1.54 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda($ Mo $K \alpha)=0.7107 \AA, \quad \mu=$ $1.233 \mathrm{~cm}^{-1}, \quad F(000)=460.0, \quad T=290 \mathrm{~K}, \quad R=0.049$ for 2047 unique observed reflexions. Bond distances © 1992 International Union of Crystallography


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[^1]:    * Lists of structure factors, anisotropic thermal parameters, H -atom parameters, torsion angles and least-squares planes have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54809 ( 12 pp .). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England. [CIF reference: AL0508]

